

# A STUDY ON CF<sub>3</sub>I-Ar AND CF<sub>3</sub>I-Kr MIXTURE GASES SUBSTITUTING SF<sub>6</sub> IN HIGH VOLTAGE EQUIPMENTS

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## Abstract:

The present binary mixtures of the SF<sub>6</sub> gas with Ar and Kr gases have not been used in many industries as long-term measures for totally eliminating the potential contribution of SF<sub>6</sub> to global warming. In order to gain more insight into electron transport coefficients in mixture gases as substitutes for SF<sub>6</sub> in high voltage equipment, transport coefficients such as electron drift velocity, density-normalized longitudinal diffusion coefficient, ratio of the longitudinal diffusion coefficient to the electron mobility, Townsend first ionization coefficient, electron attachment coefficient, and density-normalized effective ionization coefficient in CF<sub>3</sub>I-Ar and CF<sub>3</sub>I-Kr mixture gases are calculated and analyzed in the wide E/N range of 0.01 – 1000 Td using a two-term approximation of the Boltzmann equation for the energy. These calculated coefficients are analyzed and compared to those in pure SF<sub>6</sub> gas. The limiting field strength values of E/N, (E/N)<sub>lim</sub>, of these mixture gases are also derived and compared with those of the pure SF<sub>6</sub> gas at different percentages of CF<sub>3</sub>I and SF<sub>6</sub>. The mixture gases of 70% CF<sub>3</sub>I with Ar and Kr have (E/N)<sub>lim</sub> values greater than those of the pure SF<sub>6</sub> gas. Therefore, these mixture gases could be considered to substitute SF<sub>6</sub> gas in high voltage equipment.

*Key words: Trifluoroiodomethane; CF<sub>3</sub>I; SF<sub>6</sub>; Boltzmann equation analysis; Electron transport coefficients; Gas mixture.*